

Cambridge International Examinations

Cambridge International General Certificate of Secondary Education

CANDIDATE NAME CENTRE			CANDIDATE	
NUMBER			NUMBER	
PHYSICS				0625/43
Paper 4 Theory	(Extended)		Oct	tober/November 2017
			1 hour 15 minutes	
Candidates ans	wer on the Questi	ion Paner		

READ THESE INSTRUCTIONS FIRST

No Additional Materials are required.

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use an HB pencil for any diagrams or graphs.

Do not use staples, paper clips, glue or correction fluid.

DO **NOT** WRITE IN ANY BARCODES.

Answer **all** questions.

Electronic calculators may be used.

You may lose marks if you do not show your working or if you do not use appropriate units.

Take the weight of 1.0 kg to be 10 N (acceleration of free fall = $10 \,\text{m/s}^2$).

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.

The syllabus is approved for use in England, Wales and Northern Ireland as a Cambridge International Level 1/Level 2 Certificate.

This document consists of 14 printed pages and 2 blank pages.



	truck accelerates uniformly along a straight, horizontal road. The m $0 \times 10^4 \text{kg}$.	ass of the truck is
(a)) The speed of the truck increases from rest to 12 m/s in 30 s.	
	Calculate	
	(i) the distance travelled by the truck during this time,	
	distance = (ii) the resultant force on the truck.	[2]
	resultant force =	[4]
(b)) To maintain a uniform acceleration, the forward force on the truck must of Explain why.	change.
		[2]
		[Total: 8]

2 Fig. 2.1 shows a measuring cylinder that contains a coloured liquid.

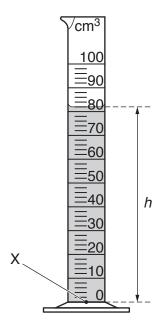


Fig. 2.1

The measuring cylinder contains 82 cm³ of the liquid. The density of the liquid is 950 kg/m³.

(a) Calculate the mass of the liquid.

mass =	Г	(J	٦
111ass –		U	1

- **(b)** The height h of the liquid in the measuring cylinder is $0.094 \,\mathrm{m}$.
 - (i) Calculate the pressure due to the liquid at point X in Fig. 2.1.

	(ii)	The true pressure at point X is different from the value calculated in (b)(i) . Explain why.
		[1]
(c)		mall object is made of steel. It is placed level with the top surface of the liquid in the suring cylinder and then released. The object sinks in this liquid.
	(i)	Explain why the object sinks in this liquid.
		[1]
	(ii)	Describe how the volume of the object can now be determined.
		[1]
		[Total: 8]

3 Fig. 3.1 shows solar cells that use radiation from the Sun to generate electricity.

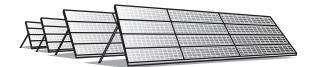


Fig. 3.1

(a)	(i)	State the name of the process which releases energy in the Sun.
		[1]
	(ii)	A reaction takes place in the Sun as energy is released.
		Describe what happens in this reaction.
		[2]
(b)	Apa from	rt from solar cells, there are other energy resources used on Earth for which the radiation the Sun is the source.
	Stat	te the name of one of these energy resources and explain whether it is renewable.
		[2]
(c)	Stat	re two advantages and two disadvantages of using solar cells to generate electricity.
	adv	antage 1
	adv	antage 2
	disa	dvantage 1
	disa	dvantage 2
		[4]

[Total: 9]

4	(a)	The molecules of most liquids are, on average, slightly further apart than the molecules of solid.
		State one other difference between the molecular structures of a solid and a liquid.
		[
	(b)	A glass tube passes through a stopper and into a glass flask. Fig. 4.1 shows that the flask completely full of a liquid and that there is also some liquid in the tube.
		stopper
		glass tube
		liquid glass flask
		Fig. 4.1
		The flask is immersed in a large beaker of very hot water. At first, the level of the liquid in thube falls, but after a short time it rises.
		(i) Explain why, at first, the level of the liquid in the tube falls.
		[:
		ii) Explain why the liquid level in the tube stops falling and starts to rise.

[Total: 6]

5 A silver spoon has a thermal capacity of 7.2J/°C.

(a)	Explain what is meant by thermal capacity.

- **(b)** The silver spoon is dropped into a saucepan of boiling water. The internal energy of the spoon increases as its temperature increases from 22 °C to 100 °C.
 - (i) Calculate the increase in the internal energy of the spoon.

	increase in internal energy =[1]
(ii)	State, in terms of the atoms, what is meant by internal energy.
	[2]

(c) The spoon is removed from the boiling water and immediately it begins to transfer energy to the surroundings. The temperature of the surroundings is 22 °C.

On Fig. 5.1, sketch a graph to show how the temperature of the spoon changes with time from the instant that it is removed from the water. [3]

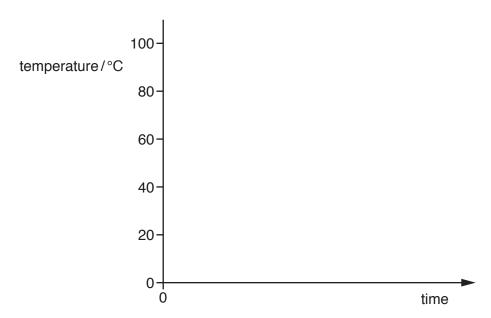


Fig. 5.1

6 Visible light is one component of the electromagnetic specified.	oectrum
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(a)	(i)	In the table, place a tick in the box next to the approximate value of the speed of	light
		in air.	[1]

speed m/s	
3.0×10^{10}	
3.0×10^{8}	
3.0×10^{6}	
3.0×10^4	
3.0×10^2	

Calculate the wavelength of this light in air.

		wavelength =[2]				
(b)	Light is travelling in an optical fibre that is made of glass.					
	(i)	The glass has a refractive index of 1.5.				
		1. Explain why the quantity refractive index does not have a unit.				
		[1]				
		2. Calculate the speed of light in the glass.				
		speed =[2]				
	(ii)	Describe one use of optical fibres in communication technology.				
		[O]				

[Total: 9]

7 (a) Fig. 7.1 shows a converging lens and its principal axis. The points F₁ and F₂ are each a principal focus of the lens.

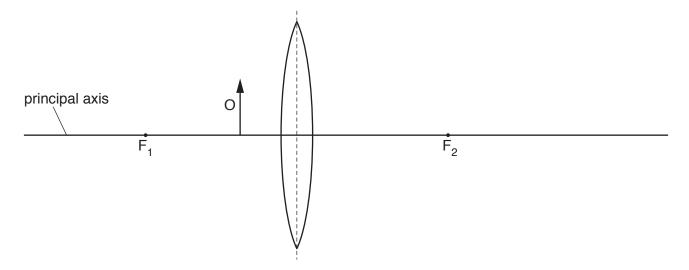


Fig. 7.1

An object O is placed between F_1 and the lens.

- (i) On Fig. 7.1, draw **two** rays from the top of the object O to locate the image.

 Label the image I. [3]
- (ii) The object O is moved to the left along the principal axis so that it is further from the lens than F_1 .

Fig. 7.2 is a diagram of the new arrangement with the new image shown.

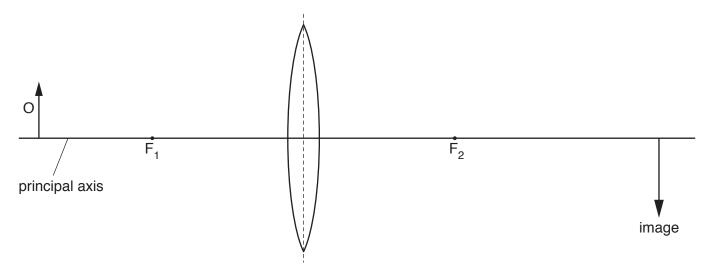


Fig. 7.2

Underline **three** of the terms below that describe the image shown in Fig. 7.2.

[2]

diminished enlarged inverted real same size upright virtual

(b) Fig. 7.3 shows yellow light passing through a glass prism.

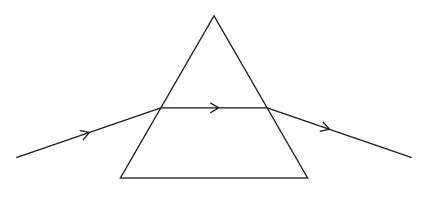


Fig. 7.3

Blue light enters the prism along the same path as the yellow light.

On Fig. 7.3, draw the path of the blue light as it enters, passes through and leaves the prism.

[2]

[Total: 7]

8 (a) Fig. 8.1 shows an uncharged conducting sphere on an insulating stand placed close to a positively charged rod.

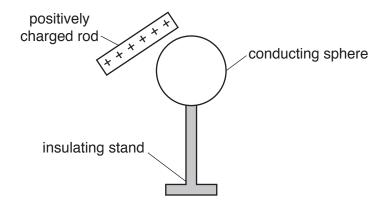


Fig. 8.1

The rod and the sphere are not moved.

sphere becomes positively charged or negatively charged.
[3
Fig. 8.2 shows a small black circle that represents a positive charge.
On Fig. 8.2, draw the pattern and the direction of the electric field in the region around the charge.

Fig. 8.2

(c) A charge of 7.0 C flows along a wire in 5.0 minutes.

Calculate the current in the wire.

(b)

current =[2]

[Total: 7]

9 (a) Fig. 9.1 shows the structure of an alternating current (a.c.) generator.

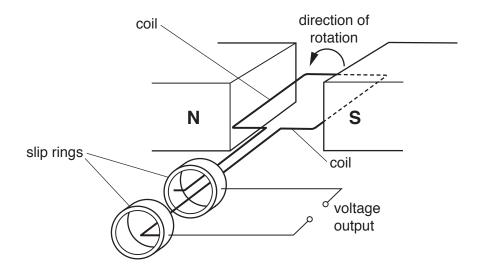


Fig. 9.1

The coil completes one rotation every 0.020 s.

(i) Using the axes in Fig. 9.2, sketch a graph to show how the voltage output of the generator varies with time during a period of 0.040 s. [2]

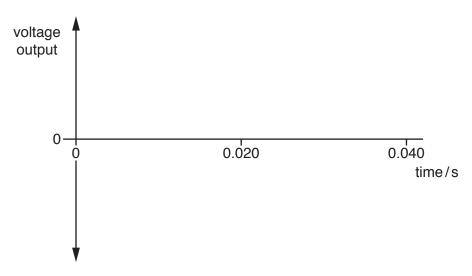


Fig. 9.2

(ii) On your graph in Fig. 9.2, mark a point labelled A to indicate a time when the coil is vertical. [1]

(b) There is an alternating current (a.c.) in a horizontal wire that is buried in a wall. A builder must miss this wire when drilling a hole in the wall.

The builder places an instrument against the wall that registers a reading when it is close to the wire. The instrument includes a long coil (solenoid) S that has an iron core and a sensitive voltmeter. Fig. 9.3 shows the circuit of the instrument close to the wire.

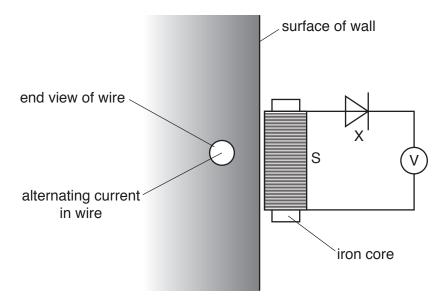


Fig. 9.3

,	Explain why there is a reading on the voltmeter.	
		[3]
(ii)	State the name and the effect of the component labelled X in Fig. 9.3.	
		[4]
		[Total: 8]

10	(a)	State the nature of γ -rays.					
			[2				
	(b)	A nucleus of technetium-99 ($^{99}_{43}$ Tc) emits only a γ -ray.					
		Stat	e any effect of this on				
		(i)	the proton number of the nucleus,				
			[1				
		(ii)	the nucleon number of the nucleus.				
			[1				
	(c)		laboratory a radiation detector displays a count rate of 16 counts/minute due to)			
		(i)	State what is meant by background radiation.				
			[1]			
		(ii)	A sample of a radioactive isotope is placed near to the radiation detector and a coun rate of 112 counts/minute is recorded.	t			
			After 18 hours, the count rate recorded is 28 counts/minute.				
			Determine the half-life of this isotope.				
	(d)		half-life =[3]			
		Radioactive isotopes are stored in thick lead containers.					
	e two precautions to be taken when radioactive isotopes are used .						
		1					
		2	[2	1			

[Total: 10]

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